

SHORT COMMUNICATIONS

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Concerning the β phase of iron(III) oxide.* By LINA BEN-DOR and ELI FISCHBEIN,† *Department of Inorganic and Analytical Chemistry* and ZVI KALMAN, *Racah Institute of Physics, Hebrew University, Jerusalem, Israel*

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β -Fe₂O₃, body centred cubic, *Ia*3, $a_0 = 9.393$ (2) Å, $Z = 16$, $D_x = 5.14$ g cm⁻³. In the 24(*d*) and 8(*b*) symmetry positions the Fe³⁺ ions are octahedrally coordinated, the former being more distorted than the latter.

Bonnevie-Svendsen (1958) was the first to announce the structure of a new cubic iron oxide phase which he prepared from the hydrolysis of FeCl₃·6H₂O. The X-ray powder diffraction pattern was similar to that of the mineral bixbyite and to β -Mn₂O₃ (also known as α -Mn₂O₃). Since α - and γ -Fe₂O₃ were already known, he assigned the letter β to the new material. The lattice constant reported by him was $a_0 = 9.40$ Å. Braun & Gallagher (1972) recently reported the structure of a new tetragonal iron oxide phase prepared from the vacuum dehydration of β -FeOOH. They, too, assign the letter β to this material having overlooked the earlier communication. Geller, Williams & Sherwood (1961) and Grant, Geller, Cape & Espinosa (1968) 'predict' the existence of the above cubic iron oxide from Vegard's Law extrapolation, having also overlooked its former discovery. Okamoto (1968), reporting the structure of a δ -FeOOH, mentions Bonnevie-Svendsen's β -Fe₂O₃, erroneously stating that one of the Fe³⁺ ions is tetrahedrally and the other octahedrally coordinated.

In the present study, a thin film‡ of Fe₂O₃ was prepared by chemical vapour deposition at 300°C, from iron triflu-

oroacetylacetone onto microscope glass, fused silica and polycrystalline alumina.

The film was polycrystalline and the diffraction pattern was taken with a Guinier camera and non-filtered Co *K* α radiation. From the X-ray data it was found that this low-temperature growth yielded a cubic iron oxide phase which was successfully indexed as β -Fe₂O₃ in accordance with Bonnevie-Svendsen's findings. On annealing at *ca* 500°C this phase transforms into α -Fe₂O₃.

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‡ Detailed growth and characterization data of the films will be published elsewhere.

References

- BONNEVIE-SVENDSEN, M. (1958). *Naturwissenschaften*, **45**, 542.
 BRAUN, H. & GALLAGHER, K. J. (1972). *Nature, Lond.* **240**, 13–14.
 GELLER, S., WILLIAMS, H. J. & SHERWOOD, R. C. (1961). *J. Chem. Phys.* **35**, 1908–1909.
 GRANT, R. W., GELLER, S., CAPE, J. A. & ESPINOSA, G. P. (1968). *Phys. Rev.* **175**, 686–695.
 OKAMOTO, S. (1968). *J. Amer. Ceram. Soc.* **51**, 594–599.

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